

**Exercise 2:** Polymer Structure in Solution and Molecular Weight Determination

**2.1:** Consider an ideal polyethylene chain with molar mass  $m = 10^7 \text{ g mol}^{-1}$ . The mean-square end-to-end distance is given by

$$\langle h^2 \rangle = Nb^2 C_\infty$$

Assume the monomer length  $b = 2.5 \text{ \AA}$ , the coefficient  $C_\infty = 6.0$ , and the molar mass of the monomer  $m_{\text{mon}} = 28 \text{ g mol}^{-1}$ .

Estimate the root-mean-square end-to-end distance  $\langle h^2 \rangle^{1/2}$  for the polymer chain.

What is the maximum length  $R_{\text{max}} = bN$  of the same polyethylene chain?

**2.2:** An NMR experiment was performed on a solution of poly(vinyl alcohol) in water at  $25^\circ\text{C}$  to determine the polymer's diffusion coefficient. A value of  $D = 1.22 \times 10^{-10} \text{ m}^2/\text{s}$  is found. Knowing that the viscosity of water at this temperature is  $8.90 \times 10^{-4} \text{ kg m}^{-1}\text{s}^{-1}$ , calculate  $R_H$ .

**2.3:** The following data were obtained for different solutions of a polymer in a membrane osmometer:

<u>c (g/L)</u>	<u>height difference measured (cm of solvent)</u>
3.2	0.70
6.6	1.82
10.0	3.10
14.0	5.44
19.0	9.3

The temperature is  $25^\circ\text{C}$  and the solvent density is  $0.85 \text{ g/ml}$ . Determine the molecular weight and second Virial coefficient of this polymer solution. How could you get the theta temperature for this polymer solvent system?

**2.4:** In an Ubbelohde-viscosimeter the viscosity of a polymer solution liquid was measured. The average time for the solution to pass the capillary in the device was 100 seconds at  $20^\circ\text{C}$ . The capillary in the viscosimeter has a diameter of  $0.8 \text{ mm}$  and the volume that passed through the capillary during the measurement was  $10 \text{ cm}^3$ . Calculate the kinematic viscosity of the solution! What additional information would you need to calculate the dynamic viscosity of the solution?

**2.5:** The following data were obtained for polystyrene solutions in butanone at  $25^\circ\text{C}$  in viscosity measurements in an Ubbelohde viscosimeter

<u>conc (g/dL)</u>	<u>t (sec)</u>
0	65.8
0.54	101.2
1.08	144.3
1.62	194.6
2.16	257.0

Determine the intrinsic viscosity and the viscosity average molecular weight of the polymer knowing that  $K$  and  $a$  for this polymer are  $0.039 \text{ cm}^3 \cdot \text{mol}^{1/2}/\text{g}^{3/2}$  and  $0.58$  respectively.